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**75336** Changes in pH and Mehlich-3 Extractable Nutrients of Selected Soils From the Midwest and South USA As Influenced by Different Rates of Iron Calcium Silicate Slag.

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Calcium silicate ( $\text{CaSiO}_3$ ) slag is among the sources of silicon (Si) in crop production and commonly obtained as by-products in steel industry and production of elemental P. These slag by-products have high calcium carbonate equivalent (CCE) making them also suitable as liming material. In the USA,  $\text{CaSiO}_3$  slag is commonly applied to soil cultivated for rice and sugarcane production at quantities ranging between 1 MT to as high as 6  $\text{MT ha}^{-1}$ . With high application rates and high CCE value of  $\text{CaSiO}_3$  slag, the solubility of essential nutrients such as P, Zn, Fe and Mn can be altered. This study was conducted to quantify the effect of different rates of iron and steel  $\text{CaSiO}_3$  slag by-product (17% Si, 81% CCE) on soil pH and extractable elements of soils collected from the Midwest and South USA. Soils were acidic (pH 5.0) to slightly alkaline (pH 7.4) with wide ranges of Mehlich-3 (M3) extractable macro- and micro-nutrients. Bulk soil samples were processed and placed in 2-kg capacity pots where  $\text{CaSiO}_3$  slag rates of 0, 1, 2, 4, 6 and 8  $\text{MT ha}^{-1}$  were thoroughly mixed with the top 15-cm surface of the potted soil. Soil samples were collected after allowing ryegrass to grow for 3 months. The pH of several soils increased linearly with increasing rates of  $\text{CaSiO}_3$ ; these soils either had initially low pH or classified as coarse-textured soils. The highest change in pH before and after  $\text{CaSiO}_3$  application was a unit of 1.05 for soils which received 8  $\text{MT CaSiO}_3 \text{ ha}^{-1}$ . Mehlich-3 extractable Mg, Mn, S, and Ca of all the soils increased with increasing rates of  $\text{CaSiO}_3$  slag. This can be attributed to the large concentrations of these elements contained in the slag material. The steady decline of M3-extractable Fe and Ni (and Zn in some soils) may be related to the potential of  $\text{CaSiO}_3$  slag to raise soil pH, hence decreasing the solubility of these metal cations. Our results show that both liming potential and composition of  $\text{CaSiO}_3$  slag had significant effect on the amount of M3-extractable essential nutrients.